# **B-NADPH**

# $\beta$ -Nicotinamide-adenine dinucleotide phosphate, reduced form (tetrasodium salt)

# reduced enzymatically

# **Structure**

#### **Formula**

: C21H26N7O17P3 · Na4

# **Formula Weight**

: 745.4 (as anhydrous free acid)

: 833.4 (as tetrasodium anhydrate)

: 887.4 (as tetrasodium trihydrate)

# **Specification**

### **Purity**

Determined by Enzymatic Method (GR) ≥ 93% **Water Content** < 8% **Na Content**  $10.0 \pm 2.0\%$ **UV Spectral Analysis**  $\epsilon$  at 260 nm and pH 10  $(14.4 \pm 0.7) \times 10^3$ 

 $\epsilon$  at 340 nm and pH 10  $(6.2 \pm 0.3) \times 10^3$ 

Ratio at pH 10

 $0.43 \pm 0.01$  $A_{340}/A_{260}$ 

# **Assay Procedure**

# **I Spectrophotometric Method**

Wavelength: 340 nm, Light path length: 1 cm Pipette the following reagents into a cuvette

	a	b	С
Tris-HCl (0.1 mol/L, pH 7.5)	5.0 mL	5.0 mL	5.0 mL
GSSG (0.1 mol/L)	0.1 mL	_	0.1 mL
NADPH (0.6 mg/mL)	0.5 mL	0.5 mL	_
GR(50 U/mL)	0.1 mL	_	0.1 mL
Distilled water	0.3 mL	0.5 mL	0.8 mL

#### **II Calculation**

$$\frac{\Delta \text{ A·V·MW} \times 100}{6.2 \times 10^{3} \cdot \text{d·v·s}} \times \frac{100}{(100 - \text{S} - \text{W})} = \text{Purity of NADPH}$$

#### $\Delta A = (Ab + Ac) - Aa$

V = Total volume of reaction mixture (6.0 mL)

MW = 745.4, anhydrous free acid

 $6.2 \times 10^3$  = Molar extinction coefficient of NADPH at 340 nm (L·mol-1·cm-1)

d = Light path length (1 cm)

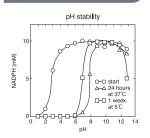
v = Sample volume (0.5 mL)

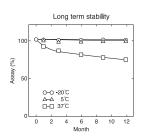
s = Sample concentration (0.6 mg/mL)

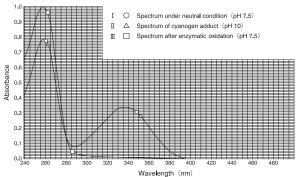
S = Na(%)

W = Water content (%)

# **Reference Data**







# Storage

Store below -20°C. Handling during short term such as transportation is allowed at 1 - 10°C. Store in the dark. Keep off humidity.

# Cat. No./Package

Cat. No.	Package	Cat. No.	Package
44330000	100 mg	44332900	Bulk
44335000	5 a		

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